

Emerging Environmental Contaminants

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U.S. Environmental Protection Agency

Office of Research & Development

Emerging Contaminants

- Sucralose (Splenda !)
- Perfluorinated Chemicals (e.g., PFOA and PFOS)
- Pharmaceuticals
- Nanomaterials
- DBPs (including NDMA and nitrosamines)
- Pesticide reaction/degradation products
- Brominated flame retardants (PBDEs)
- Perchlorate
- Algal toxins
- Benzotriazoles
- Dioxane
- Sunscreen chemicals (UV filters)
- Gasoline additives
- Naphthenic acids
- Hormones
- Pathogens

U.S. EPA



Athens, GA

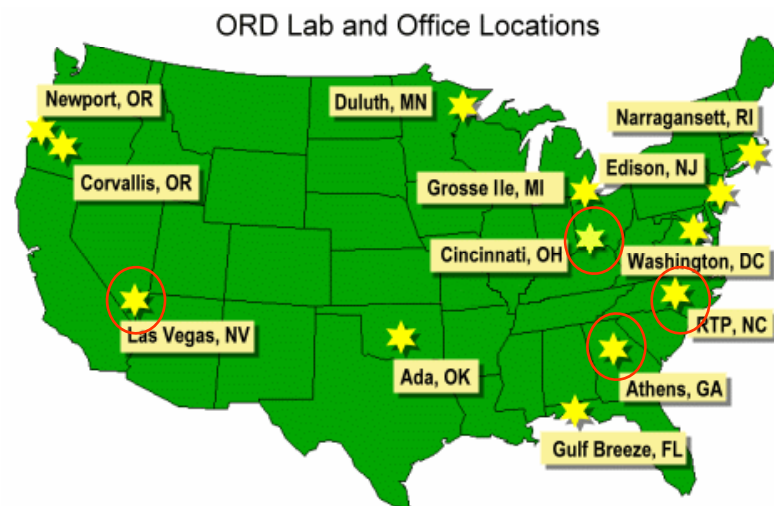


RTP, NC

Much of the emerging contaminant research described in this presentation is from EPA's Office of Research & Development's four National Exposure Research Laboratories.



Cincinnati, OH



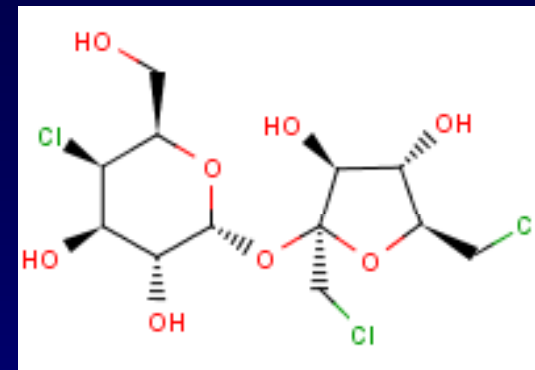
Las Vegas, NV

RESEARCH & DEVELOPMENT

Building a scientific foundation for sound environmental decisions

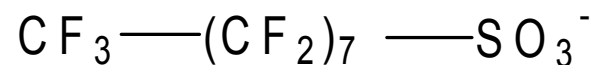
Sucralose (Splenda) !

- Artificial sweetener; widely used; very stable (can bake with it)
- Large European study (120 samples from 27 countries)
Loos et al., J. Chromatogr., A 2009, 1216, 1126-1131.
- Up to 1 µg/L in river waters; extremely persistent
- Also reported in the United States in marine and coastal waters: *Mead et al., Marine Chemistry 2009, 116, 13-17.*
- Up to 118 µg/L in wastewater effluents
Up to 372 ng/L in Cape Fear River Estuary (NC)
Up to 67 ng/L in Gulf Stream
Up to 392 ng/L in Florida Keys
- Persistent in coastal/marine waters
- Deep water levels similar to surface levels
- Significant amount is not degraded microbially; oceanic currents may distribute globally
- Other artificial sweeteners also now being measured

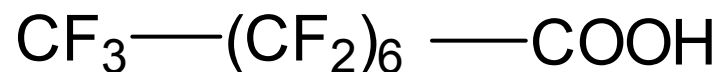


PFOS/PFOA

PFOS:
Perfluorooctanesulfonate



PFOA:
Perfluorooctanoic acid

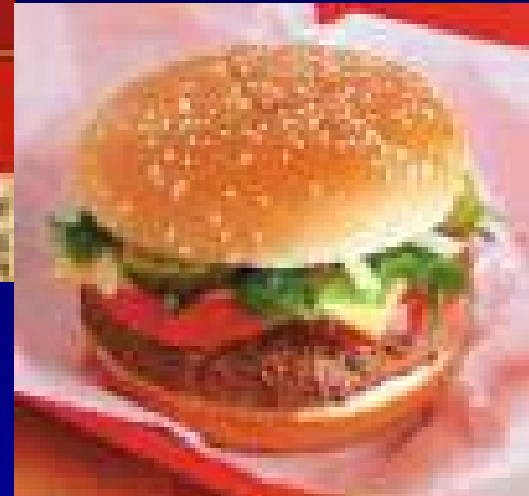


- Used to make soil-, stain-, grease-, and water-resistant coatings that are widely applied to fabrics, carpets, cookware, paper (microwave popcorn bags)
- Persistent, bioaccumulative, toxic (including developmental), ubiquitous contaminant
- **PFOA and PFOS now on new CCL-3**
- PFOS not manufactured in North America anymore;
- environmental levels decreasing
- Eight manufacturers **just phased out PFOA by 95% (2010)**



CCL-3: www.epa.gov/safewater/ccl

PFOA



Ever wonder why grease doesn't come through paper wrapper?



From Athens Sunday paper on Jan. 8

AS SEEN ON TV
TELEBrands

State of the Art Cookware

ORGREENIC™
KITCHENWARE

Green Ceramic Non-Stick Fry Pan

10" Fry Pan

Just \$19.99*
Bonus! Get 2nd Pan just pay P & H

Eco Friendly

Lifetime Warranty!

Cook Without Oil, Butter or Grease!

The future of non-stick cookware!
ORGREENIC's™ unique non-stick surface is composed of natural ceramic. This coating is Eco friendly and contains none of the harmful PFOA commonly found in non-stick cookware.

ORGREENIC's™ unique non-stick surface means you can cook & eat Healthy. It permits cooking without oil, grease or fat. Remember just 1 tablespoon of fat per day in your diet equals 10 pounds of fat in a year!

ORGREENIC's™ Ceramic coating is bonded to an aluminum alloy core. This allows for even heat and avoids hot spots so you cook evenly and efficiently. In addition, the high heat ability of the ORGREENIC™ Ceramic allows you to take your fry pan directly from the stove top to the oven anytime you want.

The Ergonomic handle diffuses heat and provides perfect weight distribution.

With the ORGREENIC™ non-stick pan you save energy, save money, save time and cook healthy. What a wonderful thing! Try the ORGREENIC™ 10" fry pan today.

Features:

- Aluminum Alloy Core
- Stay-Cool Handle
- PTFE & PFOA Free

3 Easy Ways To Order!

1. VISIT our website at: www.OrgreenicUSA.com
2. Credit card only, CALL 24 hours/7 days toll free: **1-800-945-9205**
3. Or By Mail: Send \$33.97 (*includes \$6.99 processing and handling plus \$6.99 processing for 2nd pan) to Orgreenic, Dept. #ORU0231, One Telebrands Plaza, Fairfield, NJ 07004.

TELEBrands Money Back Guarantee
Your satisfaction is 100% guaranteed. If for any reason you're dissatisfied with any merchandise, you have 30 days to return the product for an easy refund of your purchase price, excluding processing and handling.

© 2011 Telebrands. Allow 4-6 weeks for delivery.

PFOA

Questions and Concerns

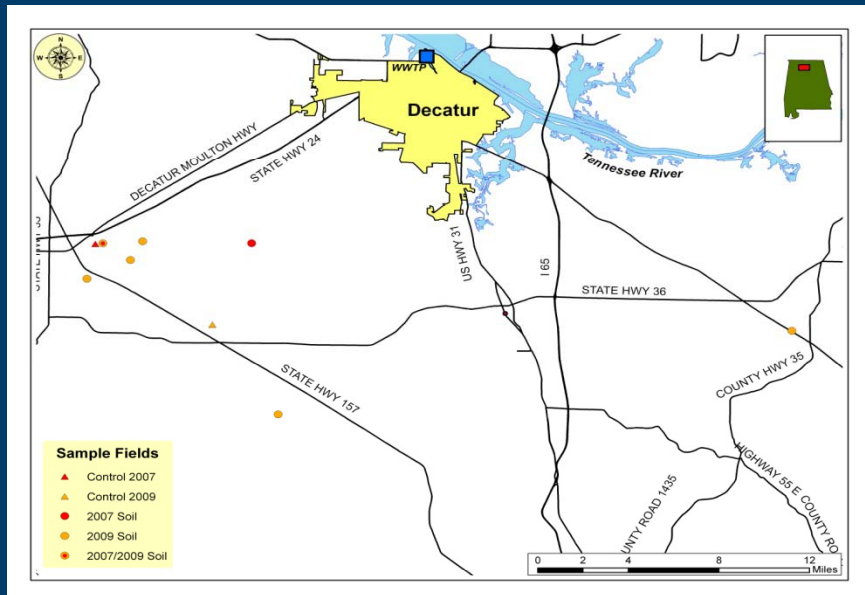
- Why is it in the bloodstream of almost everyone measured?
- Where is it coming from? How is it transported?
- Why is found in biota in the Arctic?
(Up to 3112 ng/g in polar bears from Arctic—can be at greater levels than PCBs)

Hypothesis: telomer alcohols volatile, transported, oxidized to acid in atmosphere and in the body (Henderson & Smith, *Toxicol. Sci.* 2007, 95(2), 452). Also recent evidence that PFOA itself is volatile.

Recent EPA work:

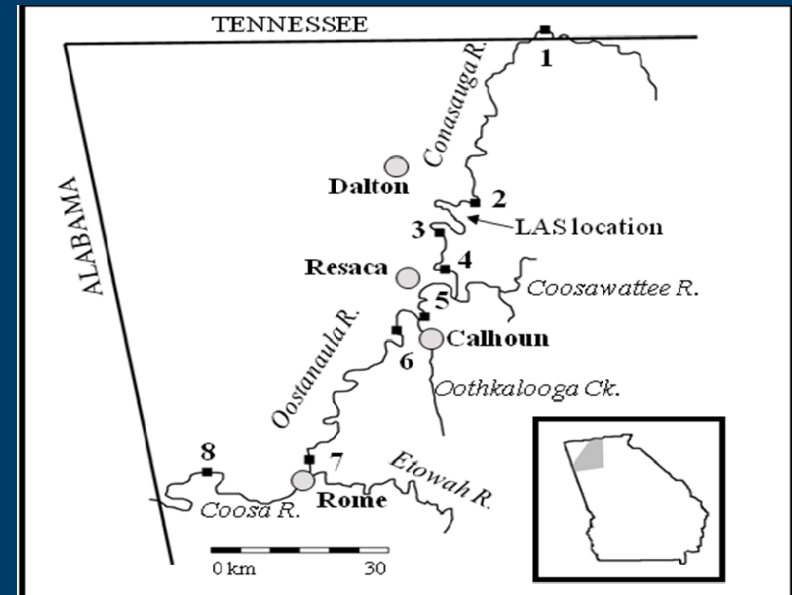
- exploring microbial degradation of fluorotelomer polymers
- investigating sources & sinks for PFOA (including sorption)
- large worldwide occurrence study in soils
- investigating source from sewage sludge used as soil improver
- measuring PFCs in fish

Measurement in Soil, Water, Sediments, Worms



Site 1: Decatur, AL

- Terrestrial
- Sludge, PFAs elevated ~50x
- Applied ~5000-acres agricultural land
- Sampled surface soil, subsurface soil,
- **Plants accumulate lower chain PFAs**
- Yoo et al., ES&T 2011, 45, 7985.



Site 2: Dalton, GA

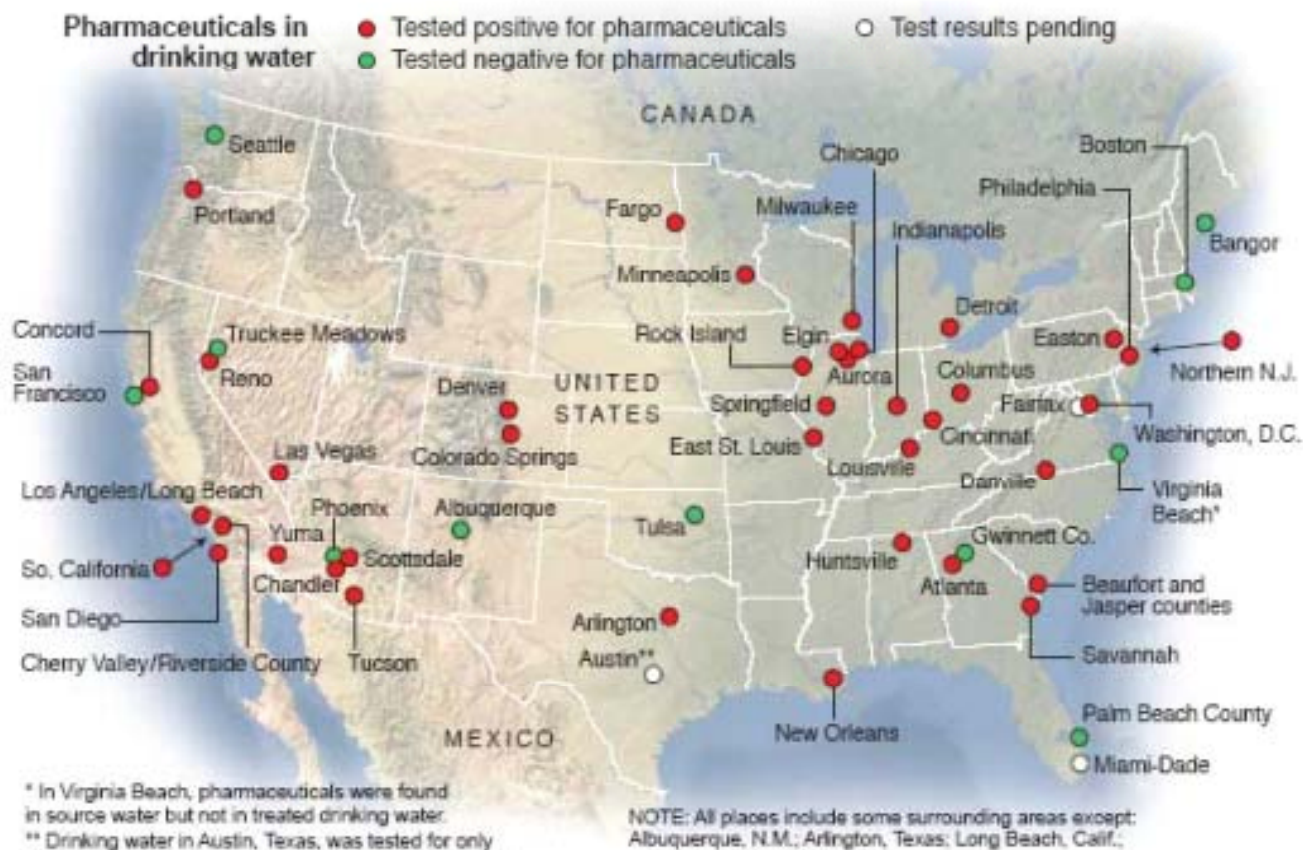
- Conasauga River
- WWTP receiving PFA waste
- Effluent irrigation near river
- Sampled water sediments – upriver & downriver
- Worms exposed to sediments in lab
- **Worms accumulate higher chain PFAs**
- Lasier et al., Environ. Toxicol. Chem. 2011, 30, 2194.

Slide courtesy of John Washington, U.S. EPA, Athens

Pharmaceuticals

46 million in U.S. have drugs in drinking water

Testing shows traces of meds in water greater than previously reported



updated 4:29 p.m. ET, Thurs., Sept. 11, 2008

Testing prompted by an Associated Press story that revealed trace amounts of pharmaceuticals in drinking water supplies has shown that more Americans are affected by the problem than previously thought — at least 46 million. That's up from 41 million people reported by the AP in March as part of an investigation into the presence of pharmaceuticals in the nation's waterways.

Pharmaceuticals

Three now included on the CCL-3; will also be included in new Water Quality Criteria

- Potential estrogenic effects on biota (e.g., feminization of fish)
- Vultures: diclofenac poisoning
- Potential antibiotic resistance
- Potential adverse effect on wastewater treatment
- May affect aquatic organism populations (decreased food sources)
- Transformation in drinking water treatment
- EPA Method 1694 (water and wastewater)





Science News

... from universities, journals, and other research organizations

Antibiotic Resistance Genes Accumulating in Lake Geneva

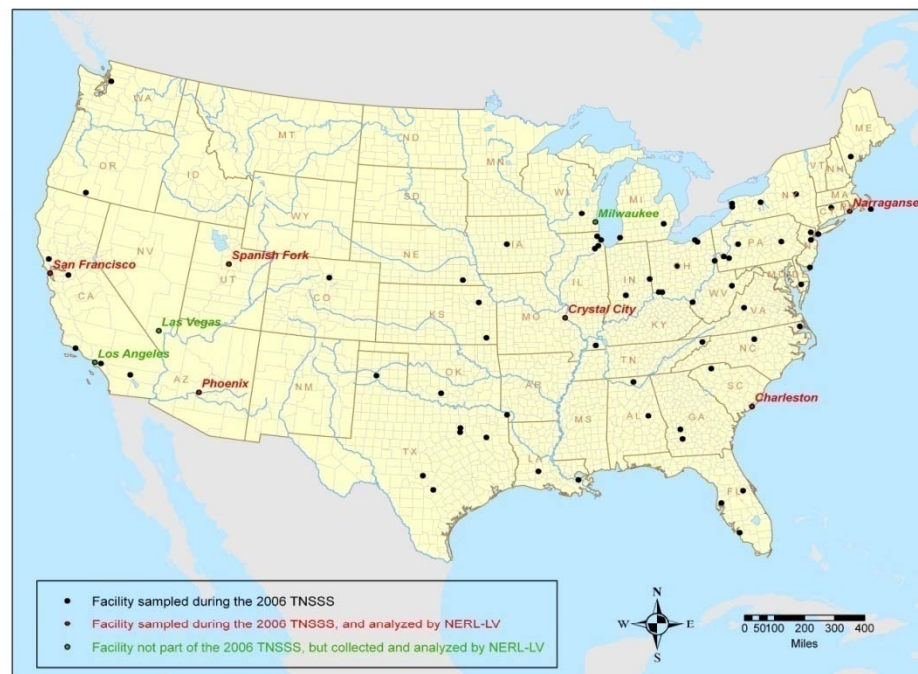
ScienceDaily (Mar. 22, 2012) — Large quantities of antibiotic-resistant bacteria enter the environment via municipal -- and especially hospital -- wastewater streams. Although wastewater treatment plants reduce the total number of bacteria, the most hazardous -- multiresistant -- strains appear to withstand or even to be promoted by treatment processes. This was demonstrated by Eawag researchers in a study carried out in Lake Geneva, near Lausanne.

Pharmaceuticals in Biosolids



Tammy Jones-Lepp

- Sampling – Samples collected from different WWTPs from across the U.S.
- Developed PLE-LC/MS/MS method to measure 6 pharmaceuticals



Results

Table 4. Concentrations of targeted pharmaceuticals and drugs from nine U.S. WWTPs using PLE 3.

Sample ID	Average amount detected [*] , ng/g (dry wt)					
	AZI	CLA	CLI	METH	n,n'-DMPEA	PSEU
LVBIO001	150	nd	nd	31	nd	nd
LVBIO002	52	nd	nd	nd	nd	nd
LVBIO003	130	310	2100	nd	nd	nd
LVBIO004	170	nd	nd	nd	630	nd
LVBIO005	200	nd	nd	nd	nd	nd
LVBIO006	53	nd	nd	nd	nd	nd
LVBIO008	180	nd	nd	nd	35	nd
LVBIO009	105	190	nd	nd	nd	140

^{*} (n=2), corrected values from matrix spikes recoveries. nd = not detected. Note: LVBIO007 was not analyzed.

AZI = azithromycin
 CLA = clarithromycin
 CLI = clindamycin
 METH = methamphetamine
 DMPEA = dimethylphenethylamine
 PSEU = pseudoephedrine

Emerging Contaminants What are They? Or a Paradigm Shift?

An Example of a Whole Lake Ethynylestradiol Dosing Experiment at Canada's Experimental Lake Area



**Jim Lazorchak, U.S. EPA,
NERL, Cincinnati, OH**

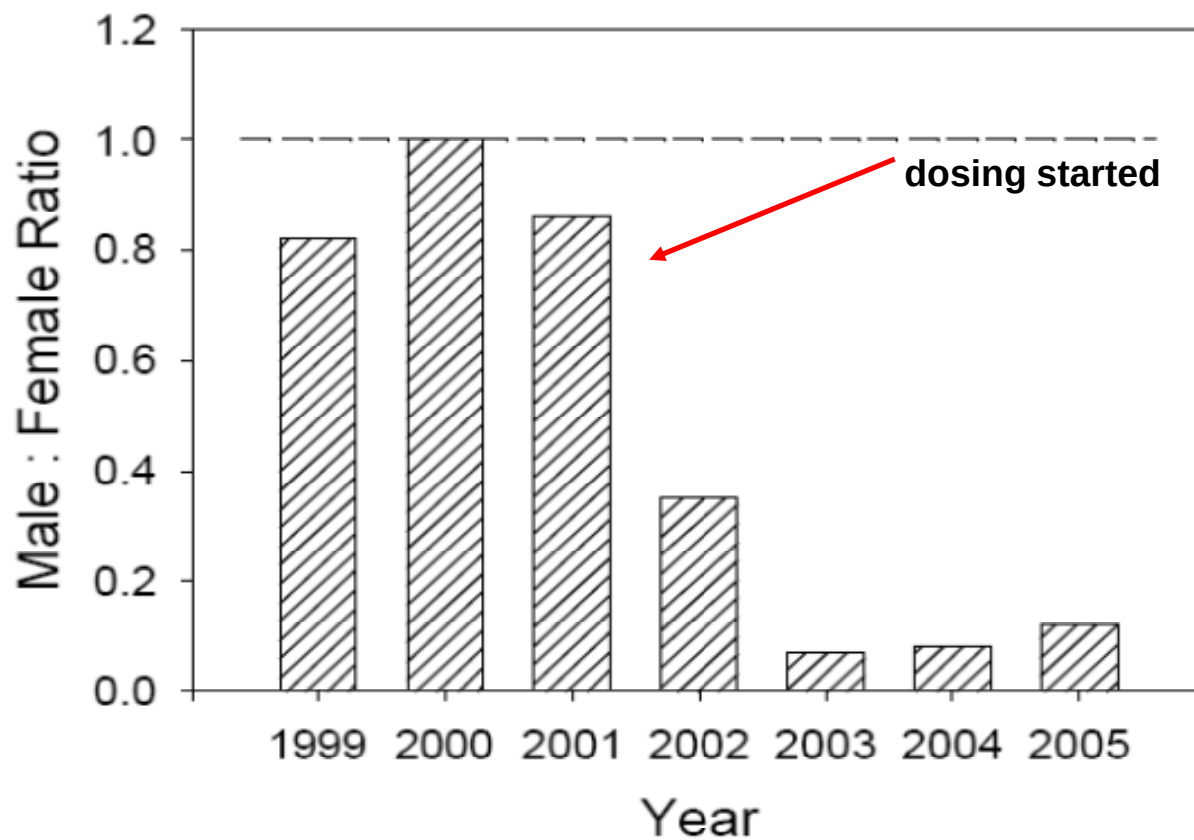
Kidd, Blanchfield, Mills, Palace, Evans, Lazorchak, Flick. 2007. Collapse of a fish population after exposure to a synthetic estrogen. *PNAS* 104, 8897-8901.

Summary - Fathead Minnow

- 
- **Spring 2001** - EE2 additions began
 - Vg gene expression induction in deployed 114 fish in 260 in 24-hrs
 - significant vitellogenin plasma induction after 7 weeks
 - **Fall 2001 (4 months)**
 - proteinaceous accumulation in kidney
 - liver cell size increased
 - **Spring 2002 (12 months)**
 - disorganized testes, immature ovaries
 - decreased spawning aggression, fewer & less-developed eggs
 - reduction in secondary sex characteristics
 - No fish population impacts observed
 - **Fall 2002 (17 months)** - reproductive failure, few age 0 fish
 - **Spring 2003 (2 years)** - only age 2 fish remaining
 - one male found, females with large ovipositors
 - **2006 (3 yrs post additions)** – Fathead population recovered



Pearl Dace Sex Ratio



K Mills – FW Inst

Nanomaterials

- Big new area for research
- Major initiative at U.S. EPA to study occurrence, fate, and health effects
- 1-100 nm in size; unique properties
- **Examples:** Fullerenes (C_{60}), zero-valent iron, TiO_2 , nano-silver, nano-gold, functionalized fullerenes (in hollow spheres, ellipsoids, tubes), quantum dots, metal oxanes
- **Already used** in many consumer products (cosmetics, sunscreens, clothing, paints, tires, tennis rackets, lubricants, electronics, soaps, shampoos, detergents; zero-valent iron used to remediate groundwater)
- Many more products under development (e.g., medical diagnostics, cancer treatment)

Nanomaterials—Ecotoxicological Effects?

- *Daphnia magna*
 - Acute toxicity (C_{60} , C-nanotubes, nano-TiO₂)
 - Accumulation (C_{60} , nano-TiO₂)
 - Physiological changes (C_{60})
 - Reproductive effects (C_{60})
 - Intake (gold nanoparticles)
- Zebra fish
 - Developmental effects, physiological changes (C_{60} , C-nanotubes)
- Rainbow trout
 - Respiratory toxicity, physiological effects (C-nanotubes)
- Frogs (*Xenopus laevis*)
 - Acute toxicity (C-nanotubes) (**but NO genotoxicity found**)
- Crustaceans (*Amphiascus tenuiremis*, *Hyalella azteca*)
 - Mortality, developmental and reproductive effects (C-nanotubes, ZnO, nanosilver)
- Various bacteria, algae
 - Acute toxicity (ZnO, nanosilver)



Farré et al., *Trends Anal. Chem.* **2008**, 27, 991-1007.

But, most studies carried out at very high doses (mg/L)

Not known how they will behave in the environment (will they aggregate and not behave as nanoparticles?)

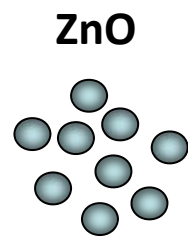
A portion of effects from metal nanomaterials may be due to the release of metal ions (e.g., silver ions)

Biotransformation and Genotoxicity of ZnO and CeO₂ Nanomaterials in Plants

XAS results

ZnO NPs
were biotransformed

-
-
-



??



Zn(OH)₂?
Zn²⁺?
ZnO



Jorge Gardea-Torresdey

But CeO₂ unchanged; both genotoxic

Photo: G. de la Rosa



López M.L., de la Rosa, G., Hernández, J.A., Michel, H., Botez, C., H. Peralta, J.R., Gardea-Torresdey, J.L.
Environmental Science and Technology. 2010, 44, 7315–7320.



Evidence for Biomagnification of Gold Nanoparticles within a Terrestrial Food Chain

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Nanoparticles from the rapidly increasing number of consumer products that contain manufactured nanomaterials are being discharged into waste streams. Increasing evidence suggests that several classes of nanomaterials may accumulate in sludge derived from wastewater treatment and ultimately in soil following land application as biosolids. Little research has been conducted to evaluate the impact of nanoparticles on terrestrial ecosystems, despite the fact that land application of biosolids from wastewater treatment will be a major pathway for the introduction of manufactured nanomaterials to the environment. To begin addressing this knowledge gap, we used the model organisms *Nicotiana tabacum* L. cv *Xanthi* and *Manduca sexta* (tobacco hornworm) to investigate plant uptake and the potential for trophic transfer of 5, 10, and 15 nm diameter gold (Au) nanoparticles (NPs). Samples were analyzed using both bulk analysis by inductively coupled plasma mass spectrometry (ICP-MS) as well as spatially resolved methods such as laser ablation inductively coupled mass spectrometry (LA-ICP-MS) and X-ray fluorescence (μ XRF). Our results demonstrate trophic transfer and biomagnification of gold nanoparticles from a primary producer to a primary consumer by mean factors of 62, 11.6, and 9.6 for the 5, 10, and 15 nm treatments, respectively. This result has important implications for risks associated with nanotechnology, including the potential for human exposure.

Introduction

Advances in nanotechnology have led to the development of many new consumer products containing nanomaterials and large increases in the mass and volume of several nanoparticles (NPs) produced. Silver (Ag) NPs have been studied extensively, due largely to their demonstrated antimicrobial effects (1). This property gives Ag NPs a wide variety of commercial applications, and their production and resultant environmental loading has increased to reflect this (2, 3). Titanium dioxide (TiO_2) and zinc oxide (ZnO) NPs are being used in products such as sunscreens, pharmaceuticals, and UV protective coatings (4, 5), and as these uses become more widespread, they will inevitably be discharged into the environment in increasing concentrations (2).

Between 2005 and 2010, the amount of listed products employing nanotechnology has increased from 54 to 1015

(3). As a result, the amount of nanomaterials entering waste streams is increasing steadily (2). During wastewater treatment, nanoparticles have been shown to concentrate in the sludge (2), 60% of which is applied to agricultural land as biosolids in the U.S. and the majority of Europe (2). Probabilistic material flow analysis models have predicted that between 2008 and 2012, the nanomaterials concentrations in sludge-treated soil in the U.S. will increase from 0.1 to 0.5 mg kg^{-1} for TiO_2 NPs, from 6.8 to 22.3 $\mu\text{g kg}^{-1}$ for ZnO NPs, and from 2.3 to 7.4 $\mu\text{g kg}^{-1}$ for Ag NPs (2).

Plant uptake (6), gastrointestinal absorption (7), and transmembrane transport of nanomaterials (8) have already been demonstrated. In the past, dietary exposure and trophic transfer have proven to be important terrestrial exposure pathways to chemicals such as polychlorinated biphenyls (PCBs), methylmercury, and dichlorodiphenyltrichloroethane (DDT) (9). Despite these facts, there has been little research published investigating the trophic transfer of nanoparticles and even less addressing the transfer from plants to animals. In one such study, trophic transfer was reported when freshwater algae (*Pseudokirchneriella subcapitata*) exposed to 10–25 nm CdSe quantum dots were fed to *Ceriodaphnia dubia* (10). Another study reported trophic transfer of 6 by 12 nm ellipsoid carboxylated and biotinylated CdSe quantum dots from ciliated protozoans to rotifers (11). A recent publication presented evidence for the transfer of 21 nm TiO_2 NPs from *Daphnia magna* to *Danio rerio* (zebrafish) (12). We have demonstrated the accumulation of Au NPs up to 55 nm in diameter (9) and copper (Cu) NPs up to 100 nm in diameter in earthworms from exposure to NP spiked soil (9, 13). There are also some studies that provide indirect evidence of trophic transfer. In one such study, researchers presented indirect evidence that gold (Au) NPs transferred from the water column to the marine food web within estuarine mesocosms (14). Another study exposed *Caenorhabditis elegans* to ZnO , Al_2O_3 , and TiO_2 NPs in the presence of *Escherichia coli* (*E. coli*), which were provided as food (15). It is possible that some of the NPs were internalized by the *E. coli* and transferred to the nematodes through dietary uptake.

These initial results suggest that NPs could be available for dietary uptake at lower trophic levels and subsequently transferred through terrestrial food webs. As terrestrial ecosystems are a possible pathway for human exposure (16), there is an urgent need to examine the transport and fate of NPs in terrestrial ecosystems under environmentally relevant scenarios. To begin addressing this knowledge gap, we used the model organisms *Nicotiana tabacum* L. cv *Xanthi* and *Manduca sexta* (tobacco hornworm) to investigate plant uptake and the potential for trophic transfer of 5, 10, and 15 nm diameter Au NPs. Au NPs are being used in a variety of applications including the detection and imaging of cancer cells (17), pharmaceuticals designed to combat HIV (18), and catalysis in fuel cells (19). We selected Au for this initial study due to its resistance to oxidative dissolution and release of dissolved Au ions as well as because of its low natural background concentrations (20). These properties make Au an ideal model to probe particle specific uptake in complex systems.

Based on previously published results (6, 7, 9, 10, 12, 13), we hypothesized that 5–15 nm Au NPs may be transferred through dietary exposure from a primary producer to a primary consumer. We also hypothesize that due to the presence of size selective physiological barriers, such as the plant cell wall, size will be an important factor in plant uptake and potential trophic transfer of Au nanoparticles. Finally,

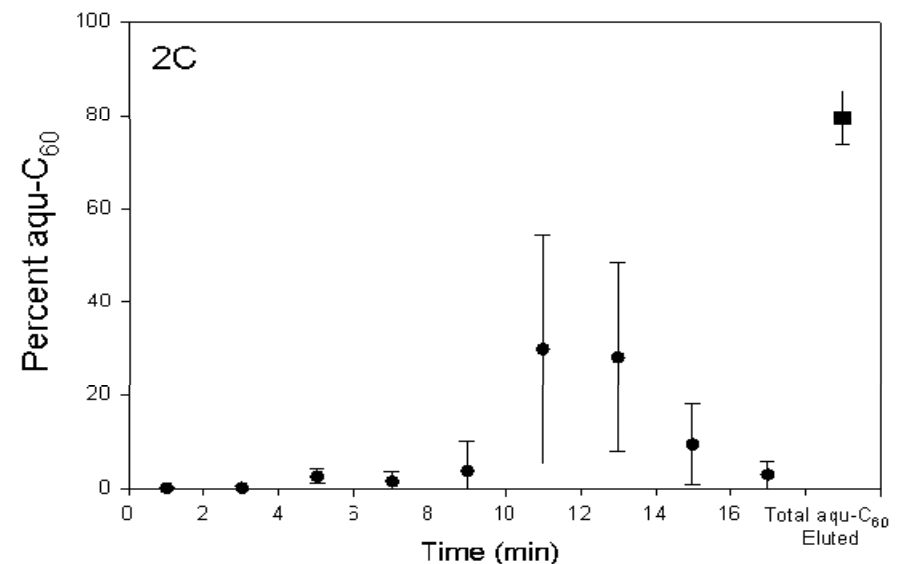
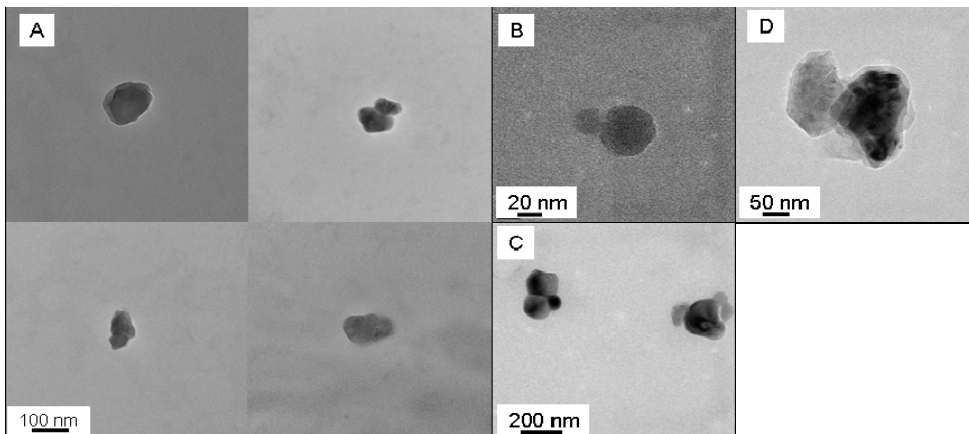
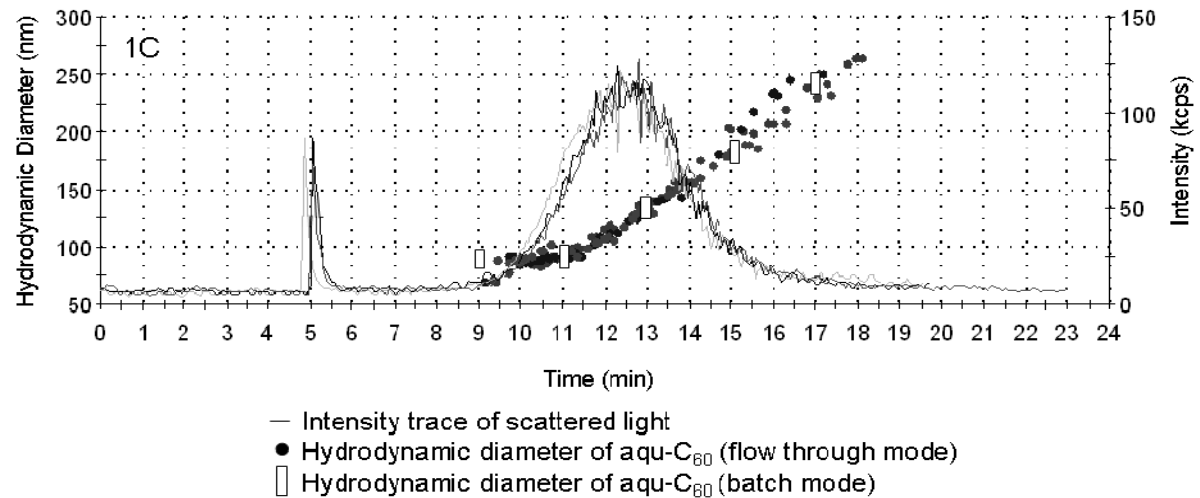
New evidence for biomagnification



* Corresponding author phone: 059/257-1651; e-mail: pmbertsch@engr.uky.edu. Corresponding author address: University of Kentucky, Department of Plant and Soil Sciences, N-212M, Agricultural Science Center North, Lexington, KY 40546.

Characterization of Fullerenes with Asymmetric Flow Field-Flow Fractionation (AF4) and LC/APPI-MS

- Aqu/C₆₀ aggregate ranged in size from 80-260 nm
- Mass of C₆₀ determined in each fraction by LC/APPI-MS
- Mass balance of $77 \pm 5.8 \%$
- Sizes corroborated by dynamic light scattering in batch mode and transmission electron microscopy
- Use of LC/MS important for specificity—to know which nanoparticles are being measured



Drinking Water DBPs—What are the Issues?

Concern over possible human health risk:

- Epidemiologic studies: **risk of bladder cancer**; some cause cancer in laboratory animals
- Recent concerns about possible **reproductive & developmental effects** (from epi studies)



Goal: Comprehensively identify DBPs formed from different disinfectants, test for toxicity, understand their formation, minimize or eliminate in drinking water

DBPs Regulated by the U.S. EPA

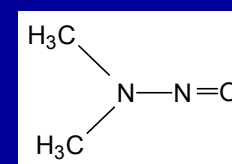
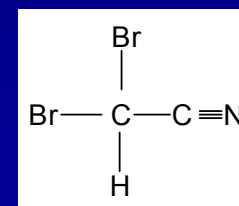
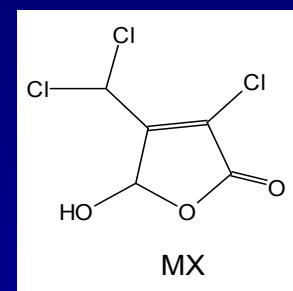
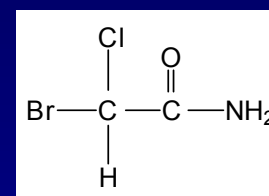
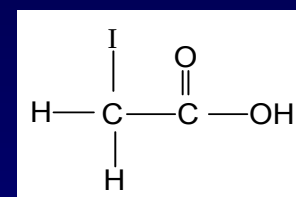
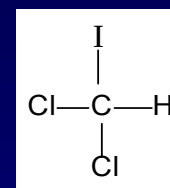
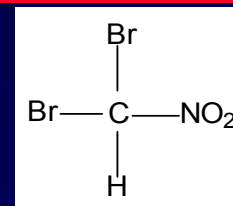
DBP	MCL ($\mu\text{g/L}$)
Total THMs	80
5 Haloacetic acids	60
Bromate	10
Chlorite	1000

But more than 600 DBPs have been identified

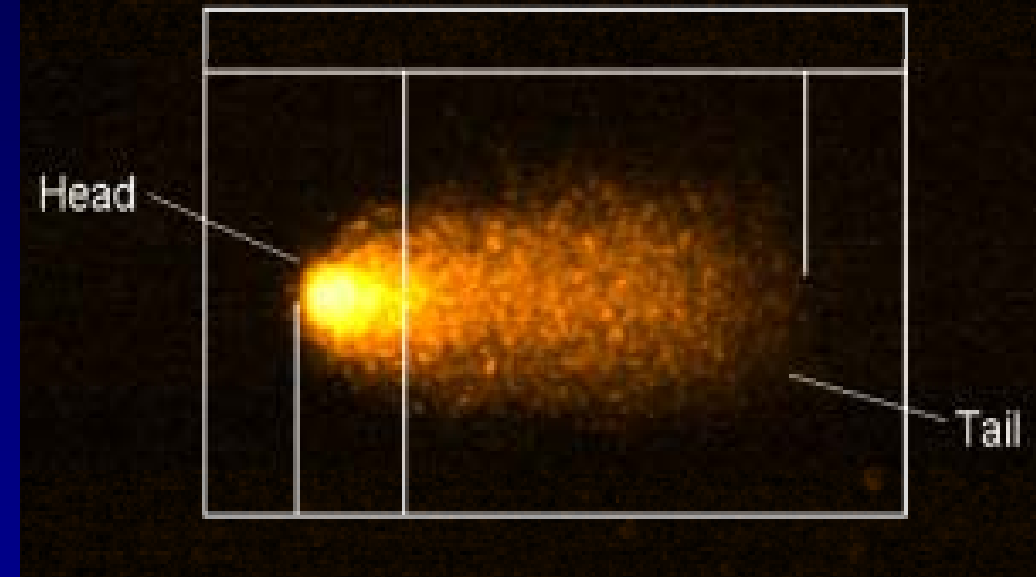
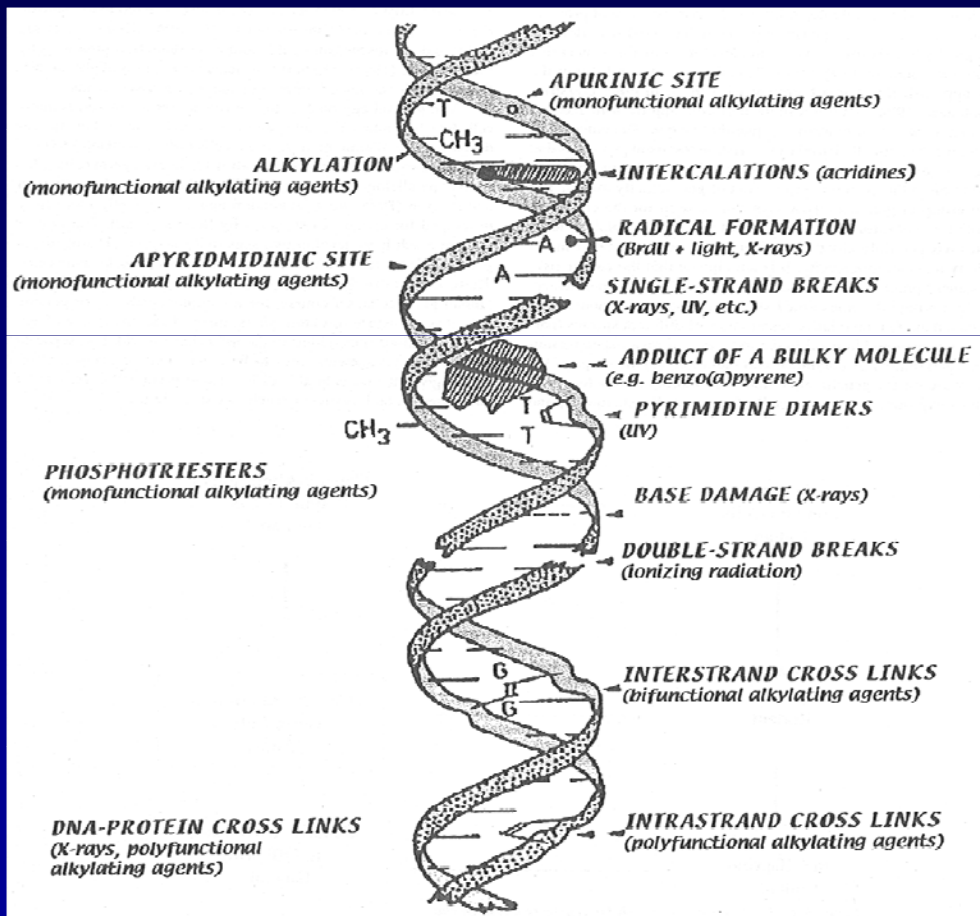
Little known about occurrence & toxicity of unregulated DBPs

Emerging DBPs

- **Halonitromethanes** (up to 3 ppb; highly genotoxic); new *in vivo* effects; increased with **preozonation**
- **Iodo-THMs and Iodo-Acids** (iodo-THMs up to 15 ppb; iodo-acids up to 1.7 ppb; both classes highly cytotoxic or genotoxic); increased with **chloramination**
Richardson et al., *ES&T*, 2008, 42, 8330.
- **Haloamides** (up to 14 ppb; highly genotoxic) may be increased with **chloramination**
- **Halofuranones** (up to 2.4 ppb for total MX analogues; genotoxic, carcinogenic); **chloramination** can also form
- **Haloacetonitriles** (up to 41 ppb; ~10% of THM4; genotoxic, cytotoxic); may be increased with **chloramination**
- **Nitrosamines** (up to 530 ppt; probable human carcinogens); increased with **chloramination**
EPA plans to regulate group of nitrosamines

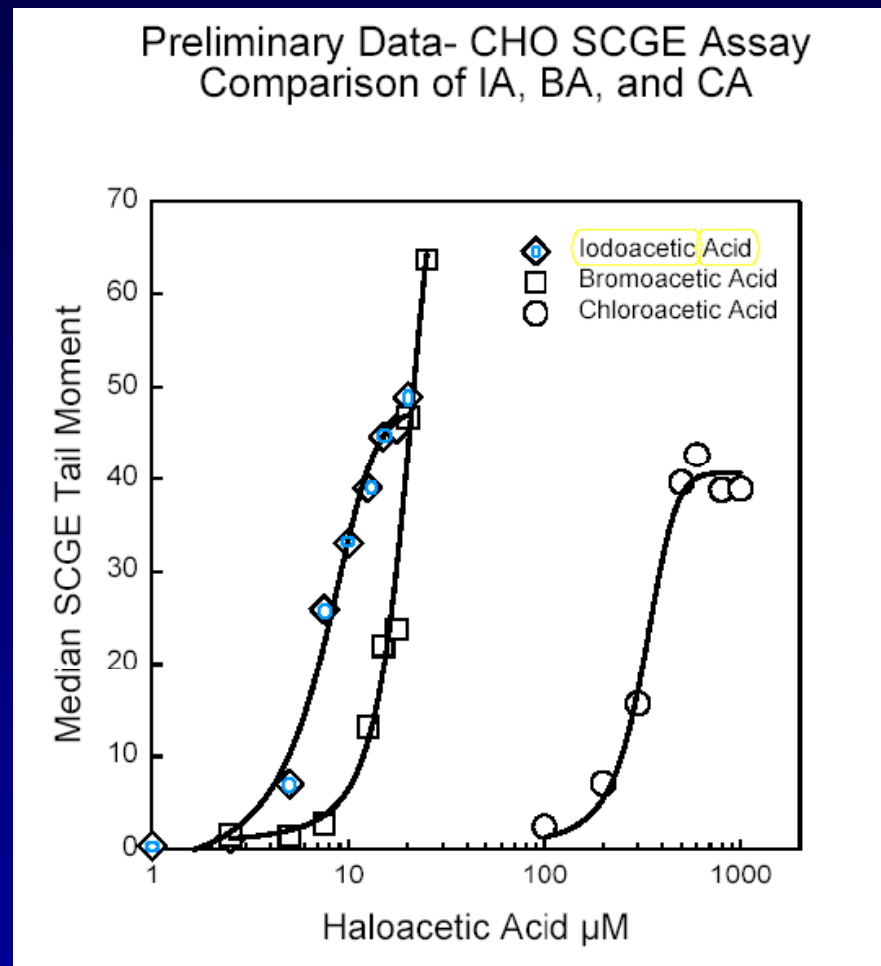


Genotoxicity: Comet Assay



The tail moment is the integrated value of DNA density multiplied by the migration distance. The % tail DNA is the amount of DNA that has migrated into the gel from the nucleus.

Genotoxicity of Iodoacetic acid

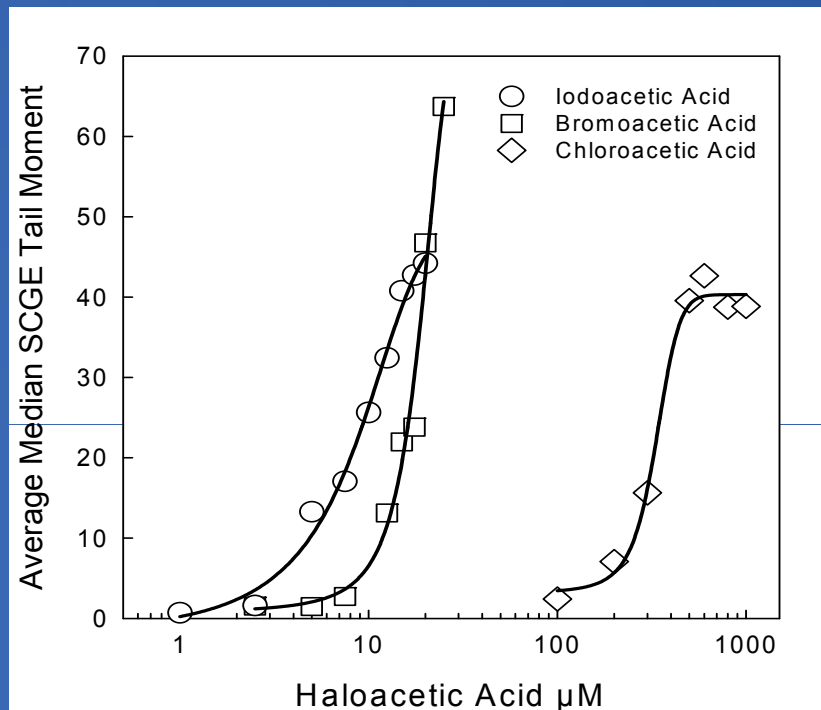


Plewa et al., *Environ. Sci. Technol.* 2004

IA also caused developmental effects in mouse embryos (Hunter et al., 1995)

Iodo-acid/iodo-THM occurrence study: Richardson et al., *ES&T*, 2008, 42, 8330.
Also reports genotoxicity & cytotoxicity of other iodo-acids and iodo-THMs

New Human Cell Results



CHO Cells

Plewa et al., ES&T. 2004, 38, (18), 4713-4722.

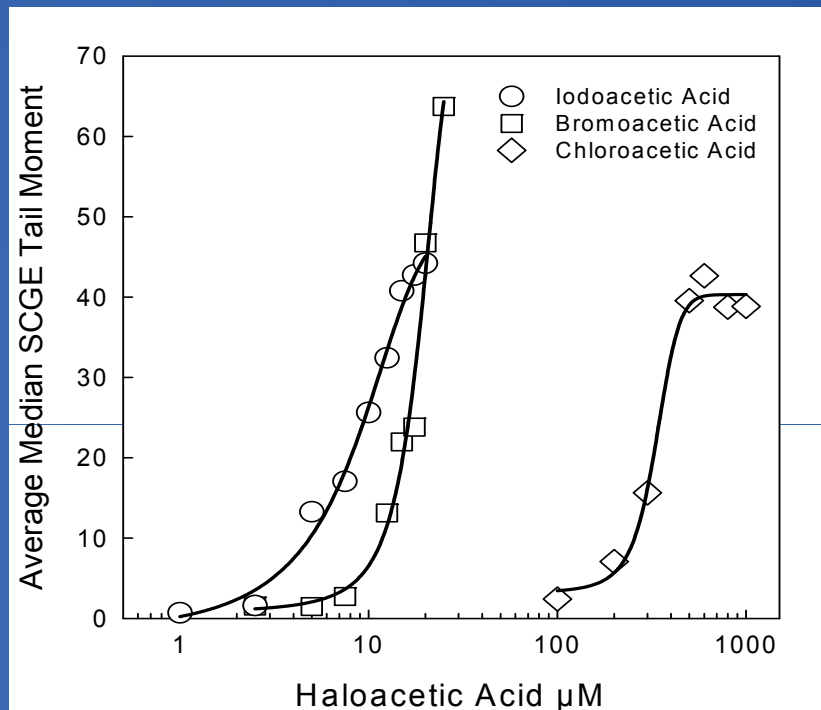


Slide courtesy of Michael Plewa, Univ. IL



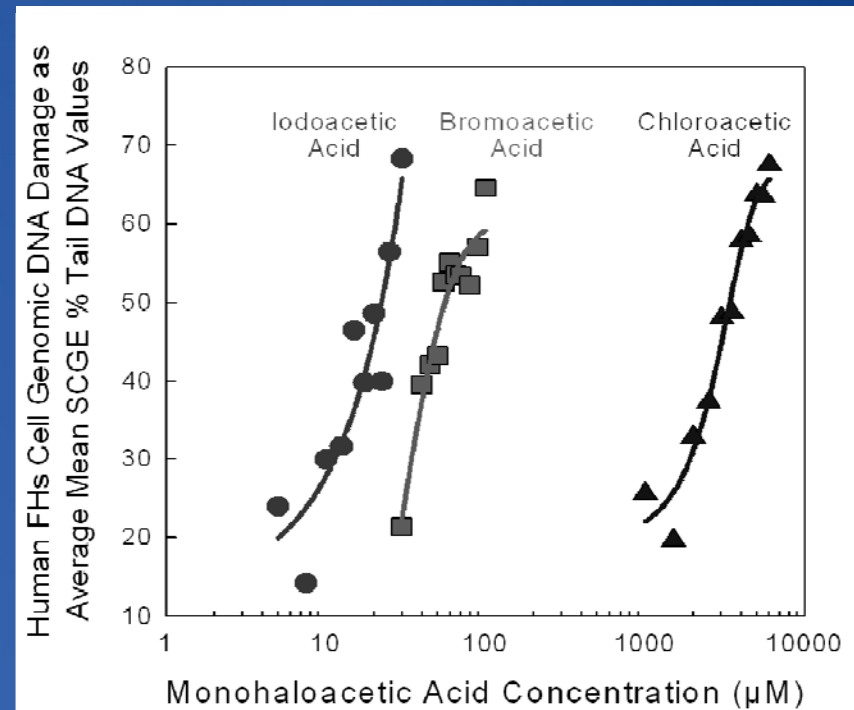
waterCAMPus

New Human Cell Results



CHO Cells

Plewa et al., ES&T. 2004, 38, (18), 4713-4722.



Human Cells

Attene-Ramos, Wagner, Plewa, ES&T, 2010, 44 (19), 7206-7212.

Nontransformed human small intestine epithelial cells



Slide courtesy of Michael Plewa, Univ. IL



waterCAMPus

Four Lab Study

Integrated Disinfection By-products Mixtures Research: Toxicological and Chemical Evaluation of Alternative Disinfection Treatment Scenarios

A collaborative effort between:

NHEERL (National Health and Environmental Effects Research Laboratory), RTP

NERL (National Exposure Research Laboratory), Athens

NRMRL (National Risk Management Research Laboratory), Cincinnati

NCEA (National Center for Environmental Assessment), Cincinnati

Purpose:

To address concerns related to potential health effects from exposure to DBPs that cannot be addressed directly from toxicological studies of individual DBPs or simple DBP mixtures

1st phase: Published in *J. Toxicol. Environ. Health* 2008 (series of 7 papers)

2nd phase: Published in *Environ. Sci. Technol.* 2010 (Chemistry paper)

Planning follow-up 4-Lab Chloramination Study



Four Lab Study: Linking DBPs measured to overall toxicity effects

In vitro:

- Reproductive/developmental
- Mutagenicity
- Carcinogenicity
- Neurotoxicity
- Metabolism



In vivo:

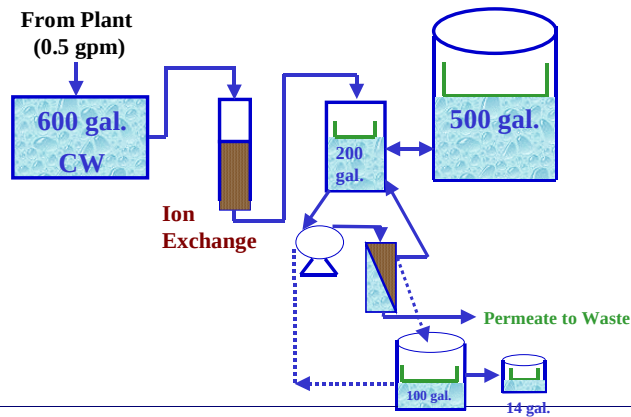
- Reproductive/developmental
- Mutagenicity/carcinogenicity
- Immunotoxicity
- Hepatic/renal toxicity
- Neurotoxicity/developmental neurotoxicity
- Kinetics/metabolism

Chemistry: 70 Priority DBPs quantified and >100 DBPs comprehensively identified using GC/MS



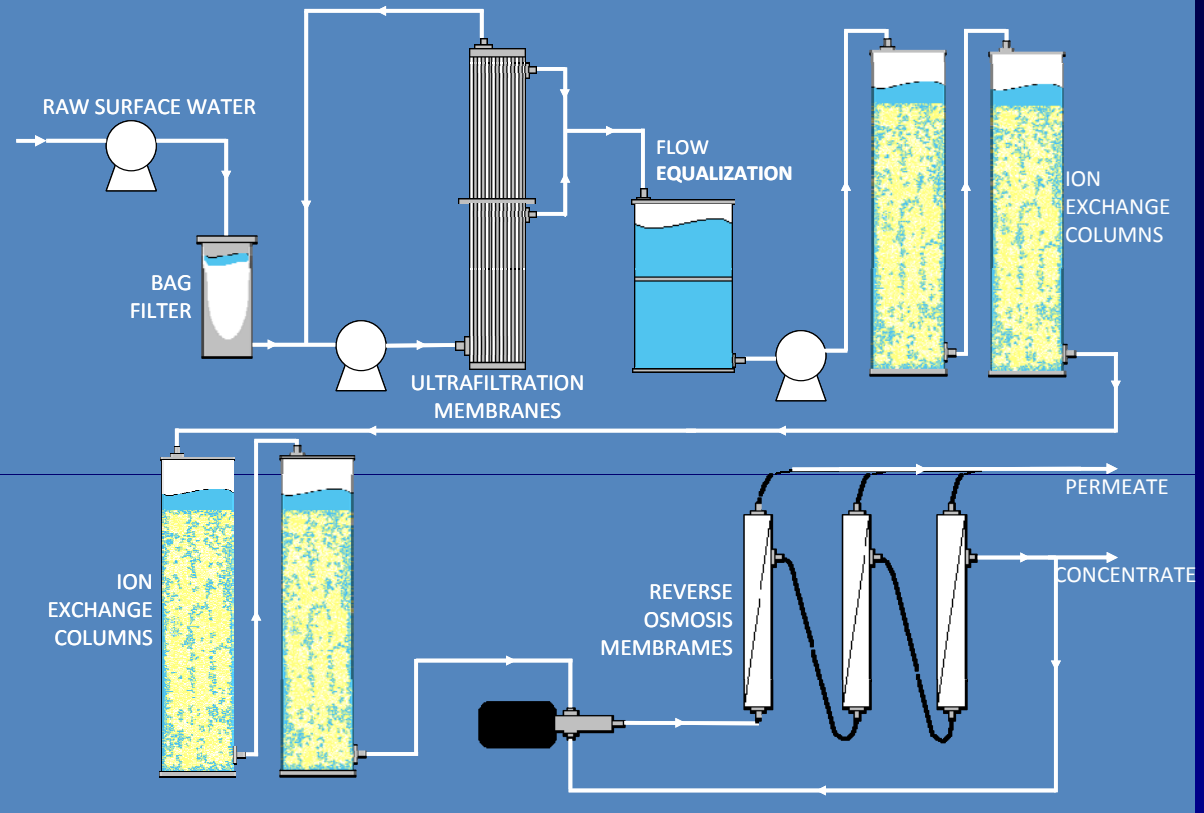
Jane Ellen Simmons

RO Concentration of DBPs



1st Phase:
 Cl_2 & $\text{O}_3\text{-Cl}_2$; treated water first
 Concentrated after

Full Study Concentration



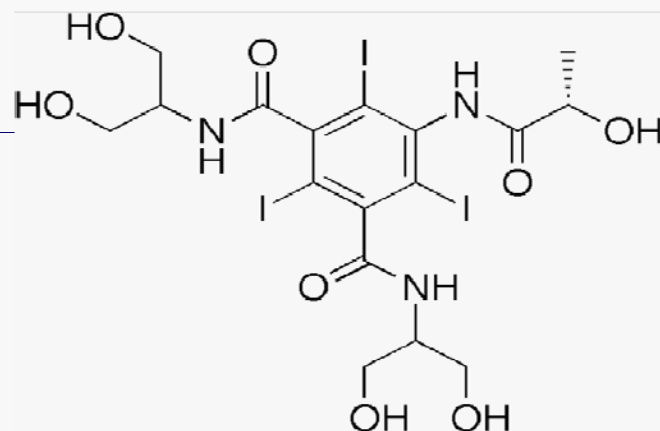
2nd Phase:
 Concentrated NOM first
 Treated with Cl_2 after

Rats can't drink organic solvents!

DBPs can also form from pollutants...

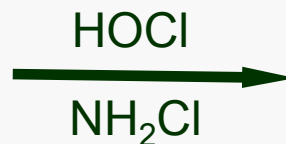
- Pharmaceuticals
- Pesticides
- Antibacterial agents
- Estrogens
- Textile dyes
- Bisphenol A
- Parabens
- Alkylphenol ethoxylate surfactants
- Algal toxins

Formation of iodo-DBPs from X-ray contrast media



Iopamidol

+ NOM



Iodo-DBPs



Iodo-DBP Occurrence Study

	Iodide ($\mu\text{g/L}$)	Sum iodo-acids ($\mu\text{g/L}$)	Sum iodo-THMs ($\mu\text{g/L}$)
Plant 2	1.0	0.37	4.9
Plant 4	ND	0.10	1.2
Plant 11	1.5	0.21	2.3
Plant 15	ND	0.17	2.4

Detection limit = 0.13 $\mu\text{g/L}$

Richardson et al., *Environ. Sci. Technol.* 2008, 42, 8330-8338.

ICM in U.S. Drinking Water Sources (ng/L)

	lopamidol	lomeprol	lopromide	lohexol	Diatrizoate
Plant 1	11	ND	ND	ND	ND
Plant 2	510	ND	24	120	93
Plant 4	110	ND	6	49	ND
Plant 10	ND	ND	ND	ND	ND
Plant 11	100	ND	ND	85	ND
Plant 12	280	ND	ND	120	ND
Plant 13	ND	ND	ND	ND	ND
Plant 15	2700	ND	25	ND	ND
Plant 17	ND	ND	ND	ND	ND
Plant 19	ND	ND	ND	ND	ND

Courtesy of Thomas Ternes, Federal Institute of Hydrology, Germany
 ICM measured using LC/ESI-MS/MS; DLs = 5-20 ng/L

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Controlled Laboratory Reactions

Experiments

- React ICM with HOCl, NH_2Cl (with and without NOM)
- 3 pHs
- Follow formation of iodo-DBPs
- Identify reaction products and intermediates
- Measure genotoxicity

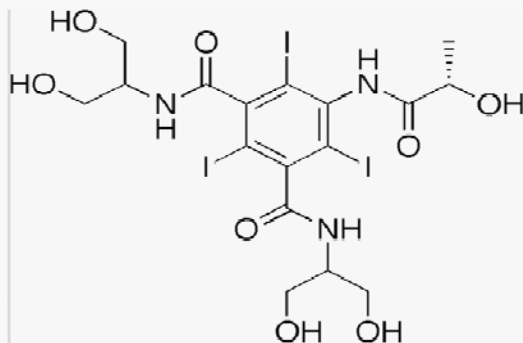
Methods

- Iodo-THMs: GC/EI-MS
- Iodo-Acids: GC/NCI-MS (with derivatization)
- Iopamidol (and other ICM): LC, LC/MS/MS
- Larger MW products and intermediates: LC/MS/MS
- Genotoxicity: Chinese hamster ovary cells, single cell gel electrophoresis

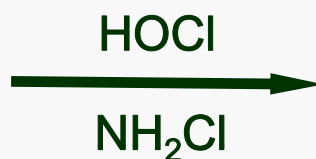


Cristal and Steve

Results

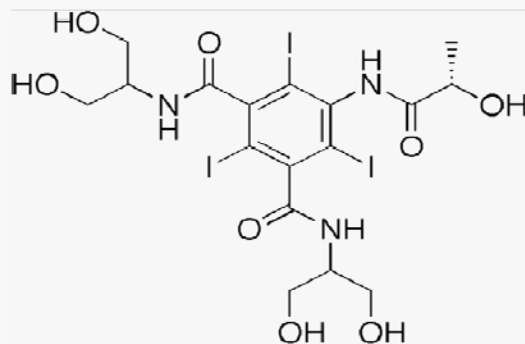


lopamidol



Iodo-DBPs

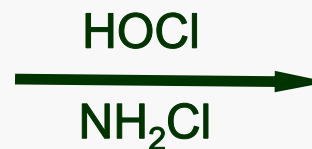
trace



lopamidol

+

NOM

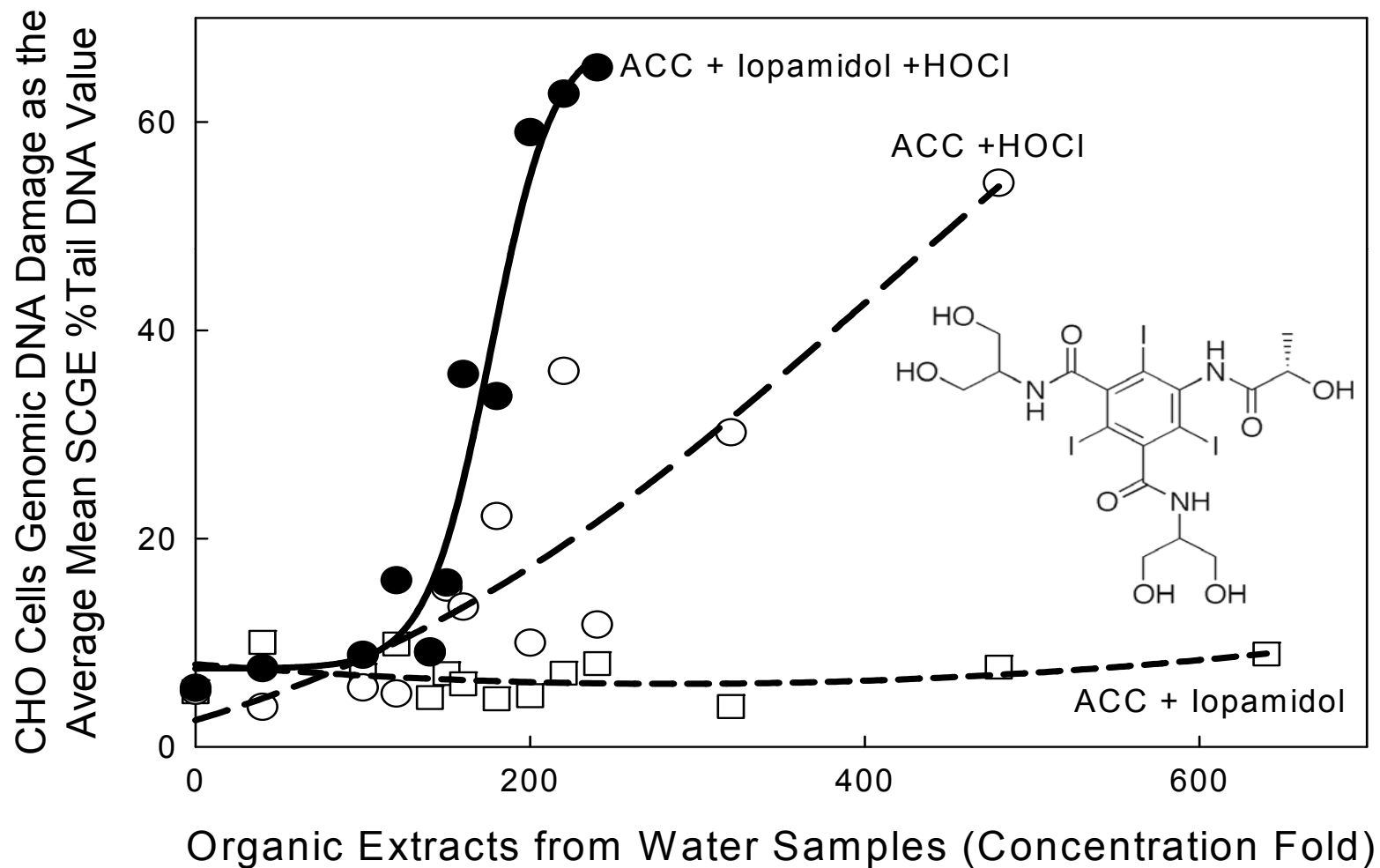


Iodo-DBPs



Iodo-THMs
&
Iodo-Acids

Genotoxicity of Chlorinated Waters Containing Iopamidol



In closing....

Ever wonder what happens when you have to
scale things up for toxicity testing???

The Land of Extraordinarily Large Lab Equipment



Chris

Toxicity?
 $20\text{ L} \rightarrow 1\text{ mL}$



Steve



Cristal

Acknowledgments

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Tom Jenkins, Matthew Henderson, Carl Isaacson
(U.S. EPA, Athens)



Thomas Ternes



Michael Plewa



Jim Lazorchak



Tammy Jones-Lepp



Jorge Gardea-Torresdey