

Entered
7-30-09
AM

STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL PROTECTION
SPRINGS SAMPLING FIELD SHEET

ation Information

STATION ID: <u>9714</u>	DATE: <u>04/09/2009</u> (MM/DD/YYYY)	TIME (EST): <u>14:19</u> (24 hr) (Final field measurements; start of sampling)
SPRING NAME: <u>Mission</u>	VENT ID: _____	
WEATHER CONDITIONS (Circle One) ☒ Clear	☐ Partly Cloudy	☐ Cloudy ☐ Rain

Water Quality Measurements

Secchi Depth: _____ (meters)	Secchi Visible on Bottom	Yes	No (Circle One)				
Stage (if available): _____ (Feet)							
Estimated depth in vent from which laboratory samples were collected: _____ (meters)							
Analytes collected for this site are as listed per the container inventory, including QA/QC samples ☒ Yes ☐ No							
Field Measurements (if field measurement result has to be qualified, enter F next to the value below):							
Estimated Depth (meters)	Meter Id	Measure ment #	pH (s.u.)	D.O. (mg/L)	% D.O.	Temperature (°C)	Conductivity (uS/cm)
	<u>QD02228-</u> <u>QT02027</u>	<u>5</u>	<u>7.5</u>	<u>0.29</u>	<u>3.3</u>	<u>22.2</u>	<u>323</u>

Sample Preservation

Was a new 0.45-micron capsule filter used to collect the final metals and nutrients/anions samples?	☒ Yes	No (Circle One)		
Were all appropriate samples preserved within 15 minutes?	☒ Yes	No		
Were total nutrient samples preserved with sulfuric acid?	☒ Yes	No	Total volume added <u>1</u> (mL)	Final pH <u>4.2</u> (s.u.)
Were filtered nutrient samples preserved with sulfuric acid?	☐ Yes	No	Total volume added <u>1</u> (mL)	Final pH _____ (s.u.)
Were total metal samples preserved with nitric acid?	☒ Yes	No	Total volume added <u>1</u> (mL)	Final pH <u>4.2</u> (s.u.)
Were filtered metal samples preserved with nitric acid?	☐ Yes	No	Total volume added _____ (mL)	Final pH _____ (s.u.)
(If additional bottles for the nutrient or metal samples were preserved, note volume of acid added and final pH for each one in the Comments)				
Were all appropriate samples, in accordance with the container inventory, put on wet ice at 4°C?	☒ Yes	No		

COMMENTS:	Conc.	Conc.
	Sulfuric Acid	Nitric Acid
	SA-827311	NA 815012
	EXP	EXP
	9/29/09	5/29/09

QA/QC TAKEN? No

TIME COLLECTED _____
(Circle One)
EQUIPMENT BLANK
FIELD BLANK
Blanks collected in accordance with Springs Initiative Monitoring SOP

Samplers Signatures (all team members present):

Laura Reese

[Signature]

Place
Barcode
Label
Here

STATION ID: 9714DATE: 4-9-09TIME (EST): 14:19

CHEMICAL STABILITY MONITORING

#	TIME (00:00)	pH (s.u.)	TEMP. (°C)	SP. COND (uS/cm)	DO (mg/L)	% SAT DO (%)	COMMENTS
1	14:11	7.47	22.34	322	0.34	3.8	Begin @ 14:00
2	14:13	7.46	22.30	322	0.32	3.6	
3	14:15	7.46	22.30	322	0.30	3.5	
4	14:17	7.46	22.22	323	0.29	3.3	
5	14:19	7.47	22.23	323	0.29	3.3	
6							
7							
8							
9							
10							
11							
12							
13							
14							
15							

Meter Id: QD02228-Q+02027Field Measurements Analyzed By: PM (initials)Field Measurements Recorded By: PM (initials)

The final three measurements will be evaluated for each of the field parameters listed above (only the values in mg/L for DO) to ensure that representative measurements have been collected for a site. The lowest and highest values for the three measurements must be within 20% when calculating the relative percent difference (RPD). If these values are not within 20%, the final field measurement value will be qualified as being estimated with an F value qualifier.

$$\text{RPD} = \frac{(\text{Highest Value} - \text{Lowest Value})}{(\text{Mean Value of Highest and Lowest Values})} \times 100\%$$