

STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL PROTECTION
SPRINGS SAMPLING FIELD SHEET

*Final
8-10-09
HJ*

RQ-2009-04-06-3

RQ-2009-05-11-16

Station Information

STATION ID: 9707 DATE: 5-13-09 TIME (EST): 15:41 (24 hr)
(MM/DD/YYYY) (Final field measurements; start of sampling)

SPRING NAME: Chassahowitzka Main VENT ID: _____

WEATHER CONDITIONS (Circle One) Clear ☐ Partly Cloudy ☐ Cloudy ☐ Rain ☐

Water Quality Measurements

Secchi Depth: 3.8 L (meters) Secchi Visible on Bottom ☒ Yes ☐ No (Circle One)

Stage (if available): _____ (Feet)

Estimated depth in vent from which laboratory samples were collected: 3.8 (meters)

Analytes collected for this site are as listed per the container inventory, including QA/QC samples ☒ Yes ☐ No

Field Measurements (if field measurement result has to be qualified, enter F next to the value below):

Estimated Depth (meters)	Meter Id	Measurement #	pH (s.u.)	D.O. (mg/L)	Temperature (°C)	Conductivity (uS/cm)	% DO
3.8	QD02228- Q102027	9	7.2	3.90	24.2	8830	48.0

Sample Preservation

Was a new 0.45-micron capsule filter used to collect the final metals and nutrients/anions samples? Yes ☐ No ☒ (Circle One)

Were all appropriate samples preserved within 15 minutes? ☒ Yes ☐ No

Were total nutrient samples preserved with sulfuric acid? ☒ Yes ☐ No Total volume added 2 (mL) Final pH 2 (s.u.)

Were filtered nutrient samples preserved with sulfuric acid? Yes ☐ No ☐ Total volume added _____ (mL) Final pH _____ (s.u.)

Were total metal samples preserved with nitric acid? ☒ Yes ☐ No Total volume added 2 (mL) Final pH 2 (s.u.)

Were filtered metal samples preserved with nitric acid? Yes ☐ No ☐ Total volume added _____ (mL) Final pH _____ (s.u.)

(If additional bottles for the nutrient or metal samples were preserved, note volume of acid added and final pH for each one in the Comments)

Were all appropriate samples, in accordance with the container inventory, put on wet ice at 4°C? ☒ Yes ☐ No

COMMENTS:	Sulfuric Acid	Nitric Acid
- algae pieces	SA 903501	NA 903502
- floating in boil	EX1 2/04/10	EXP
- Water at surface	- 2 mL of acids	2/04/10
- above spring was cloudy	added to Nitric + Sulfuric bottles; 1st mL added	
	didn't show pH change so	
	second mL added	

QA/QC TAKEN? ☒ NO

TIME COLLECTED _____
 (Circle One)

EQUIPMENT BLANK ☐

FIELD BLANK ☐

Blanks collected in accordance with
 Serious Initiative Monitoring SOP



MAY 13 2009

Samplers Signatures (all team members present):

Laura [Signature] *[Signature]*

CHASSAHOWITZKA MAIN

SPR0904 1ST MAG. SPRING

15:41

STATION ID: 9707DATE: 5-13-09TIME (EST): 15:41

CHEMICAL STABILITY MONITORING

#	TIME (00:00)	pH (s.u.)	TEMP. (°C)	SP. COND (uS/cm)	DO (mg/L)	% SAT DO (%)	COMMENTS
1	15:25	7.15	24.15	8850	^{4.22} 7.15	52.0	Began purging @ 15:20
2	15:27	7.16	24.09	8840	4.14	50.9	
3	15:29	7.16	24.11	8840	4.08	50.2	
4	15:31	7.17	24.15	8840	4.03	49.5	
5	15:33	7.18	24.17	8840	3.98	49.0	
6	15:35	7.18	24.23	8840	3.94	48.6	
7	15:37	7.19	24.29	8830	3.92	48.4	
8	15:39	7.19	24.18	8830	3.91	48.1	
9	15:41	7.20	24.23	8830	3.90	48.0	
10							
11							
12							
13							
14							
15							

Meter Id: QD02228-Q102007 Field Measurements Analyzed By: UH (initials)Field Measurements Recorded By: UH (initials)

The final three measurements will be evaluated for each of the field parameters listed above (only the values in mg/L for DO) to ensure that representative measurements have been collected for a site. The lowest and highest values for the three measurements must be within 20% when calculating the relative percent difference (RPD). If these values are not within 20%, the final field measurement value will be qualified as being estimated with an F value qualifier.

$$RPD = \frac{(\text{Highest Value} - \text{Lowest Value})}{\text{Mean of Highest and Lowest Values}} \times 100\%$$