

STATE OF FLORIDA
DEPARTMENT OF ENVIRONMENTAL PROTECTION
SPRINGS SAMPLING FIELD SHEET

Entered
2-2-12
5-29-12

RA-2011-09-05-48

on Information

STATION ID: 11479 DATE: 8/31/11 TIME (EST): 13:33 (24 hr)
(MM/DD/YYYY) (Final field measurements; start of sampling)
SPRING NAME: Marion Salt VENT ID: _____
WEATHER CONDITIONS (Circle One) Clear Partly Cloudy Cloudy Rain

Water Quality Measurements

Secchi Depth: 4.5 (meters) Secchi Visible on Bottom Yes No (Circle One)
Stage (if available): _____ (Feet)
Estimated depth in vent from which laboratory samples were collected: 4.5 (meters)
Analytes collected for this site are as listed per the container inventory, including QA/QC samples Yes No
Field Measurements (if field measurement result has to be qualified, enter F next to the value below):

Estimated Depth (meters)	Meter Id	Measurement #	pH (s.u.)	D.O. (mg/L)	% D.O.	Temperature (°C)	Conductivity (uS/cm)
<u>4.5</u>	<u>QD03629</u>	<u>7</u>	<u>7.4</u>	<u>2.63</u>	<u>31.8</u>	<u>23.9</u>	<u>4700</u>

Sample Preservation

Is a new 0.45-micron capsule filter used to collect the final metals and nutrients/anions samples? Yes No (Circle One)
Are all appropriate samples preserved within 15 minutes? Yes No
Are total nutrient samples preserved with sulfuric acid? Yes No Total volume added 1 (mL) Final pH 5.2 (s.u.)
Are filtered nutrient samples preserved with sulfuric acid? Yes No Total volume added _____ (mL) Final pH _____ (s.u.)
Are total metal samples preserved with nitric acid? Yes No Total volume added 1 (mL) Final pH 5.0 (s.u.)
Are filtered metal samples preserved with nitric acid? Yes No Total volume added _____ (mL) Final pH _____ (s.u.)
For additional bottles for the nutrient or metal samples were preserved, note volume of acid added and final pH for each one in the Comments)
Are all appropriate samples, in accordance with the container inventory, put on wet ice at 4°C? Yes No

Comments:

H₂SO₄ #106111 exp: 6/2/12
HNO₃ #1110070 exp: 6/23/11



QA/QC BLANK 2

SPRG1107

12:15

QA/QC TAKEN? Yes

TIME COLLECTED 12:15

(Circle One)

EQUIPMENT BLANK

FIELD BLANK

Blanks collected in accordance with



13:33

MARION SALT SPRINGS

SPRG1107 2ND MAG. SPRING 8/31/11

Signatures (all team members present):

Laura

Dhy

STATION ID: 11479DATE: 8/31/11TIME (EST): 13:33

CHEMICAL STABILITY MONITORING

#	TIME (00:00)	pH (s.u.)	TEMP. (°C)	SP. COND (uS/cm)	DO (mg/L)	% SAT DO (%)	COMMENTS
1	13:19	7.42	23.40	4710	2.70	32.6	Begin purging @ 13:09
2	13:23	7.41	23.97	4700	2.68	32.4	
3	13:25	7.41	23.92	4690	2.68	32.3	
4	13:27	7.41	23.92	4700	2.68	32.1	
5	13:29	7.41	23.84	4710	2.64	31.8	
6	13:31	7.40	23.85	4700	2.64	31.6	
7	13:33	7.40	23.89	4700	2.63	31.8	
8							
9							
10							
11							
12							
13							
14							
15							

Meter Id: QD03629Field Measurements Analyzed By: UH (initials)Field Measurements Recorded By: UH (initials)

The final three measurements will be evaluated for each of the field parameters listed above (only the values in mg/L for DO) to ensure that representative measurements have been collected for a site. The lowest and highest values for the three measurements must be within 20% when calculating the relative percent difference (RPD). If these values are not within 20%, the final field measurement value will be qualified as being estimated with an F value qualifier.

$$\text{RPD} = \frac{(\text{Highest Value} - \text{Lowest Value})}{(\text{Mean Value of Highest and Lowest Values})} \times 100\%$$