HOT BLOCK DIGESTION OF SOIL, SEDIMENT, WASTE AND TISSUE SAMPLES FOR TOTAL RECOVERABLE METALS

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1. SCOPE AND APPLICATION

1.1. This method details the hot block digestion of soil, sediment, tissue, and waste samples for total recoverable metals analysis. This method is based on EPA method 3050B.

1.2. Samples digested by this method are suitable for analysis by Inductively Coupled Plasma Atomic Emission Spectroscopy (ICP-AES) (DEP SOP MT-061) and Inductively Coupled Plasma Mass Spectrometry (ICP-MS) (DEP SOP MT-067 and DEP SOP MT-074).

1.3. This method is not a total digestion technique for most samples. It is a very strong acid digestion that will dissolve almost all elements that could become ‘environmentally available’. By design, elements bound in silicate structures are not normally dissolved by this procedure as they are not usually mobile in the environment.

2. SUMMARY OF THE METHOD

A representative (0.5 - 1 g wet weight or approximately 0.5 g dry weight) sample is digested by refluxing with nitric acid and hydrogen peroxide. The sample is then further

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refluxed with hydrochloric acid. A separate sample portion is dried for a percent dry solids determination if necessary.

3. APPARATUS AND EQUIPMENT

3.1. Eppendorf pipette (adjustable volume, 100-1000 uL).

3.2. Finn pipette (adjustable volume, 2-10 mL).

3.3. 50 mL Environmental Express polypropylene digestion tubes.

3.4. Reflux caps.

3.5. Balance - Top-loading or analytical balance with a readability of at least 0.001 g.

3.6. Stainless steel spatulas.

3.7. Digestion block - Environmental Express SC-100 located inside a fume hood.

3.8. Aluminum and polystyrene weigh boats.

3.9. Environmental Express resin pellets solid matrix, catalog # SC400.

3.10. Digital thermometer with a NIST-traceable manufacturer's certificate of calibration.

4. REAGENTS AND STANDARDS

See DEP SOP MT-069 for further instruction on reagent documentation. All bottles of reagents and standards (purchased and prepared) in the laboratory are required to have laboratory information management system (LIMS) labels which include the unique serial number.

4.1. Deionized (DI) water.

4.2. Trace metal grade nitric acid (HNO3) - The expiration date is provided by the manufacturer. Stored in the fume hood cabinet at room temperature.

4.3. Trace metal grade hydrochloric acid (HCl) - The expiration date is provided by the manufacturer. Stored in the fume hood cabinet at room temperature.

4.4. 30 % hydrogen peroxide solution (H_2O_2) - No expiration date. Stored in the fume hood cabinet at room temperature.

4.5. 50 % HNO_3 - Prepare by diluting 500 mL HNO_3 to 1 L with DI water. Always add acid to water and stir to dissipate heat. The expiration date will be the same as for the acid used to make the dilution.

4.6. Spiking Solutions: These solutions are purchased or prepared by the metals analysis group. **Inform the supervisor of the Metals Group when the prepared stocks are either within two weeks of expiration, as indicated by the**

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label on the bottle, or within two weeks of running out. Refer to Appendix A for concentrations of the spiking solutions.

4.6.1. SED SPK 4A and SED SPK 4B - Purchased (i.e. Fisher or High Purity) single-element stock standards used to prepare the spiking solution for the 'LFBSED' laboratory fortified blank. The expiration date of these standards will be no longer than the expiration dates of the stock individual metal standards provided by the manufacturer. Stored at room temperature.

4.6.2. ICPMINA spike - Custom stock solution from Spex CertiPrep purchased through Fisher used to prepare the minerals laboratory fortified blank (LFBMIN). Expiration date provided by the manufacturer. Stored at room temperature.

5. SAMPLE COLLECTION, PRESERVATION, AND HANDLING

5.1. Soil, sediment, waste, and tissue samples for metals analysis are stored in one of the refrigerators located in the receiving room.

5.2. The holding time from time of collection to digestion is 180 days.

5.3. All glassware should be cleaned according to DEP SOP MT-007.

6. SAMPLE PREPARATION PROCEDURE

6.1. Check the samples out from the receiving room and set aside to allow equilibration to room temperature. Create a new prep worksheet in the Prep Worksheet Manager module in LIMS (this step may be done before or during the sample preparation process). The Prep ID will be 3050-SED, 3050-WAS, or 3050-TISS, depending on the matrix being digested (see section 8).

6.2. Print the required sample labels and place each one on the outside of a digestion tube. Be sure to include the quality control (QC) required for the batch (see section 8).

6.3. Sample Homogenization - If the sample cannot be thoroughly mixed within its container without spilling, empty the entire sample into a clean stainless steel tray. Do not decant any liquid or remove any objects from the sample. Please contact the supervisor if any non-representative material is found in the sample. A mortar and pestle can be used if necessary. Any large metal pieces should be cut up or crushed, but not removed from the sample. Return the homogenized sample to the jar.

6.4. Weighing Samples:

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6.4.1. Soil and Sediment samples - The sample results for soil and sediment samples are reported on a dry weight basis. Therefore, the percent dry solid results should be determined in order to calculate the proper sample wet weight needed to obtain a final dry weight of approximately 0.5 g. Refer to DEP SOP MT-055 for the determination of percent dry solids.

6.4.1.1. Place a dry sample tube onto the balance and tare the balance. Weigh into the tube the amount of homogenized sample needed to reach approximately 0.5 g dry weight. Record the weight on the prep sheet to the nearest 0.001 g in the field marked 'sample weight'. Make sure that the entire sample is at the bottom of the vessel and that none has dropped onto the balance pan outside the vessel.

6.4.2. Waste and Tissue samples - The sample results for waste and tissue samples are reported on a wet weight basis. Place a dry sample tube onto the balance and tare the balance. Weigh into the tube approximately 0.5 g of homogenized sample. Make sure that the entire sample is at the bottom of the vessel and that none has dropped onto the balance pan outside the vessel. Record the weight on the prep sheet to the nearest 0.001 g.

6.5. Under the fume hood, **SLOWLY** add 5 mL of 50 % HNO_3 to each vessel. Beware of a possible violent reaction with certain matrices. If foaming occurs, gently swirl the sample tubes to break up the bubbles. Samples that exhibit a high reactivity with the acid should be allowed to sit for approximately 30 minutes before heating.

6.6. Preheat the hot block to approximately 95 ºC (temperature reading of a digestion tube, not the hot block display). The temperature of the heating block must be monitored and recorded before digestion on each day of use. Place 1 tube containing DI water randomly on the block and insert a thermometer. After the temperature has stabilized and reached the acceptable temperature range of 92 – 97 °C, read the temperature of the thermometer inserted into the digestion tube and record it on the LIMS prep worksheet in Prep Worksheet Manager in the 'Block Temperature' box. The thermometer used in this temperature verification must be calibrated against a NIST traceable thermometer annually. To check the calibration of a thermometer, use the procedure from DEP SOP LB-035, Calibration of Lab Thermometers using a Constant Temperature Calibration Bath. The thermometer calibration should be recorded in the 'Lab Thermometer Calibration Logbook (Fluke 7103 Temperature Calibration Microbath)' (stored in B319). Label the thermometer with the date of calibration. Alternatively, digital thermometers with manufacturer's NIST-traceable certificate of calibration may be used.

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6.7. Cover each vessel with a reflux cap and place each one into the hot block and reflux for 15 minutes without boiling. Remove the samples from the hot block and allow them to cool.

6.8. Add 2.5 mL of concentrated HNO_3 to each tube, replace the reflux caps, and put the sample tubes back into the hot block and reflux for an additional 20 minutes. Remove the tubes from the hot block and allow them to cool. For samples with high organic content, additional refluxing may be necessary. This will be evident by the generation of brown fumes upon refluxing during this step. If brown fumes evolve from the sample, repeat this step for those samples until no further brown fumes are detected. If additional refluxing is required, note this in the comments section of the prep sheet.

Note: Do not allow the solution volume to reduce low enough to let the bottom of the digestion vessel become exposed. If a digestion vessel is heated to dryness, contact the supervisor. The sample will need to be re-digested.

6.9. Continue to heat the samples at 95 °C for one hour without boiling and then remove the vessels from the hot block and allow them to cool. The reflux cap is not needed during this step; however, do not allow the samples to dry completely. It is important to ensure complete coverage of the bottom of the vessel at all times.

6.10. Add 1 mL of DI water and 1.5 mL of a 30 % H_2O_2 solution to each vessel and cover with a reflux cap. Place the vessels back into the hot block and heat (the sample solution will effervesce from the addition of the H_2O_2). Heat the samples until the effervescence subsides. Care must be taken to prevent sample loss from excessive effervescence. Remove the sample tubes from the hot block and allow them to cool. Add additional H_2O_2 in 0.5 mL aliquots followed by heating until the effervescence is minimal or the general appearance of the solution is unchanged. **Do not add more than a total of 5 mL H_2O_2 to each sample.** Contact the supervisor if 5 mL of peroxide has been added but the samples are still effervescing.

6.11. Cover each tube with a reflux cap, if not already covered, and carefully heat without boiling for 90 minutes or until the solution volume is reduced to about 2.5 mL. Periodically observe the ongoing digestion and make sure the volume in the digestion vessel is sufficient to cover the bottom of the digestion vessel. Remove the tubes from the hot block and allow them to cool. Add 5 mL of concentrated HCl to each vessel. Place onto the hot block and reflux for 15 minutes. Remove the vessels from the hot block and allow them to cool. The samples are now ready to be brought to volume.

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6.12. Reconstitute the samples to 50 mL with DI water. Cap the tubes and gently invert 8-10 times to mix the sample completely. The samples are now ready for analysis.

6.13. Write the prep worksheet (e.g. 3050-SED, WAS, or TISS, _####) ID on a label and place the label on the sample rack. Make sure all documentation is completed properly on the prep worksheet including but not limited to the prep person, digestion start and end date, digestion start and end time, reagent serial numbers, spike serial numbers, sample IDs, block ID, block temperature, and sample comments. Be sure to note any and all information in the comments section that might be useful to the analyst and/or reviewer. When the tray and prep worksheet are **completely** finished, notify a supervisor who will verify the completeness and correctness of the prep worksheet. The supervisor will then turn the samples from 'W' to 'V' status by completing the prep worksheet in Prep Worksheet Manager. This must be done BEFORE the tray is sent down to the metals lab. Print a copy of the prep worksheet to provide to the analyst with the sample tray and take them to the Metals lab, B310.

6.14. Dilutions are prepared in the analysis lab prior to sample analysis using instrument sample cups or 15 mL centrifuge tubes. Generally, a dilution of 5 is required for samples with an ICP-MS test.

7. SAMPLE ANALYSIS PROCEDURE

Samples prepared in this manner are suitable for analysis by ICP-AES or ICP-MS. See appropriate SOP for analysis, ICP-AES (DEP SOP MT-061) or ICP-MS (DEP SOP MT-067 and DEP SOP MT-074).

8. QUALITY CONTROL

8.1. A digestion batch is defined as a set of 20 samples or less consisting of samples with similar matrices and having similar physical characteristics and of the same general composition.

8.1.1. The three possible matrix groups are:

1. Soils, fresh water sediments, marine sediments, and similar materials. This group is reported on a dry weight basis. LFBs for a sediment/soil matrix digestion batch must consist of ~0.5 g of Environmental Express resin pellets, to represent a solid matrix, along with their appropriate spiking solutions. Weights must be recorded appropriately.

2. Aqueous wastes, non-aqueous wastes, solid wastes, and similar materials. This group is reported on a wet weight basis. For solid

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matrices the LFB must consist of ~0.5 g of Environmental Express resin pellets and the appropriate spiking solutions.

3. Tissues of all kinds. This group is reported on a wet weight basis. For solid matrices the LFB must consist of ~0.5 g of Environmental Express resin pellets and the appropriate spiking solutions.

8.1.2. Samples that are grouped in the same matrix group and share similar physical characteristics and general composition are to be digested in the same batch. If there are more than 20 samples of the same matrix, arrange the samples in batches of twenty samples or less to be digested together. Each batch requires its own set of quality control samples.

8.2. Selecting Samples for Quality Control - Randomly choose a sample to be used for the preparation of matrix spikes. This sample should best represent the samples in the batch.

8.3. Digestion Blank - One per digestion batch of 20 samples or less. Add ~0.5 g of Environmental Express resin pellets to the digestion blank when preparing soils, sediments, wastes, and tissues.

8.4. Lab Fortified Blanks (LFB) – The 'LFBSED' is prepared by spiking 1 mL of SED SPK 4A and 1 mL of SED SPK 4B to 50 mL of DI water. One 'LFBSED' is required for all batches. If minerals (Al, Ca, Fe, Mg, K, or Na) are requested, an 'LFBMIN' is also required. Spike 1 mL of the ICPMINA spike into 50 mL of DI water. Environmental Express resin pellets are used as the quality system matrix for all LFBs. Add approximately 0.5 g of the resin pellets to the LFB.

8.5. Duplicate Matrix Spikes – Spike 1 mL of SED SPK 4A and 1 mL of SED SPK 4B onto two separate aliquots of the QC sample. If minerals are requested, spike 1 mL of ICPMINA onto another aliquot of the QC sample.

9. SAFETY/HAZARDOUS WASTE MANAGEMENT

9.1. Gloves, safety glasses/goggles and lab coats must always be worn when handling all samples, spike solutions, acids, hydrogen peroxide, and other reagents.

9.2. Excess sample and spike solutions should be disposed of in the containers labeled Metals Waste. When 90 % full, these containers must be emptied into the large metals waste drums in the laboratory waste rooms.

9.3. Reagents should be added to samples in a fume hood with the sash lowered as far as possible to reduce exposure to fumes and protect in case of splashing.

9.4. Post analysis sample digestates - If positive results exceeding the pollutant limitations (sewer discharge limits, see DEP SOP MT-007) are found, the sample

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digestates should be discarded in the container labeled Metals Hazardous Waste**.** See the Health and Safety Plan or your supervisor if you have any questions. All other sample digestates can be disposed of in the sink as long as the acid concentration is less than 2 % or they are diluted by a factor of 50 or greater.

10. REFERENCES

10.1. EPA Method 3050B, Revision 2, December 1996

10.2. DEP SOP MT-007, Procedure for Cleaning Metals Labware

10.3. DEP SOP MT-055, Determination of Percent Dry Solids

10.4. DEP SOP MT-061, Analysis of Selected Metals by Dual View Inductively Coupled Plasma-Atomic Emission Spectroscopy (ICP-AES)

10.5. DEP SOP MT-067, Standard Operating Procedure for the Inductively Coupled Plasma Mass Spectrometer (ICP-MS) with Collision Cell Technology (CCT) and Kinetic Energy Discrimination (KED)

10.6. DEP SOP MT-069, Metals Standard and Reagent Preparation Procedure

10.7. DEP SOP MT-074, Standard Operating Procedure for the Thermo ICAP Inductively Coupled Plasma Mass Spectrometer

10.8. Health and Safety Plan, located on the DEP Bureau of Laboratories intranet.

10.9. Quality Manual, Bureau of Laboratories, DEP

10.10. DEP SOP LB-035, Calibration of Lab Thermometers using a Constant Temperature Calibration Bath.

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Appendix A Spike Concentrations

Purchased Standard Data Sheet

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Serial #:	56553
Original Serial #:	
Identity:	ICPMINA
Description:	Spike intermediate for Al, Ca, Fe, Mg, K, Na.
Type:	Purchased Solution
Group:	METALS
Status:	Active
Location:	B411
Storage Temp:	Room Temperature
Solvent Used by Manufacturer:	N/A
Prepped by Manufacturer:	
Expires:	30-JUL-2015 00:00
Entered:	21-JUL-14 by Joshua Ayres
Discarded:	
Cancelled:	
Comments:	
Lot #:	25-109CR
Source:	Spex CertiPrep
Grade:	
Purity:	
Certificate of Analysis:	Not Available
Received:	21-JUL-2014 00:00
Volume:	500 mL

Table: Purchased
Std/Reag Type: STANDARD

Component	Concentration	Mol Wt Factor
Aluminum	1000 ug/mL	1
Calcium	4000 ug/mL	1
Iron	1000 ug/mL	1
Magnesium	2000 ug/mL	1
Potassium	500 ug/mL	1
Sodium	4000 ug/mL	1

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Prepared Standard Data Sheet

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Serial #:	55610	Table: Standard	
Original Serial #:		Std/Reag Type:	
Identity:	SED SPK 4A		
Description:	SED Spk 4A use 1mL/50mL samples for spk factor of 1.0		
Type:	Working		
Group:	METALS		
Status:	Active		
Location:	B411		
Storage Temp:	Room Temperature		
Solvent Used:	Water		
Prepped:	19-MAY-2014 10:10 by AYRES_J		
Expires:	30-MAY-2015 23:59		
Entered:	21-MAY-14 by Joshua Ayres		
Discarded:			
Cancelled:			
Comments:			
Holding Time:	999 days, 0 hours		
Final Volume:	500 mL		
SOP Used for Preparation:		**See Also:**	
Verified:			
Recertified:			

Component	Parent S/N	Parent Identity:	Amt of Parent Added	Final Conc	Dilution Factor	Mol Wt Factor
Aluminum	55453	AL SECONDARY	20 mL	400 mg/L	0.04	1
Arsenic	55455	AS SECONDARY	10 mL	20 mg/L	0.02	1
Barium	55456	BA SECONDARY	2 mL	40 mg/L	0.004	1
Beryllium	55457	BE SECONDARY	.1 mL	.2 mg/L	0.0002	1
BORON	55458	B SECONDARY	10 mL	20 mg/L	0.02	1
Cd	55459	CD SECONDARY	.5 mL	1 mg/L	0.001	1
Calcium	55460	CA SECONDARY	20 mL	400 mg/L	0.04	1
CHROMIUM	55461	CR SECONDARY	1 mL	20 mg/L	0.002	1
COBALT	55462	CO SECONDARY	5 mL	10 mg/L	0.01	1
COPPER	55463	CU SECONDARY	.25 mL	5 mg/L	0.0005	1
Iron	55464	FE SECONDARY	20 mL	400 mg/L	0.04	1
Lead	55465	PB SECONDARY	1 mL	20 mg/L	0.002	1
Magnesium	55466	MG SECONDARY	20 mL	400 mg/L	0.04	1
Manganese	55467	MN SECONDARY	1 mL	20 mg/L	0.002	1
Molybdenum	55468	MO SECONDARY	5 mL	10 mg/L	0.01	1
NICKEL	55469	NI SECONDARY	5 mL	10 mg/L	0.01	1
Potassium	55470	K SECONDARY	20 mL	400 mg/L	0.04	1
Selenium	55471	SE SECONDARY	10 mL	20 mg/L	0.02	1
Sodium	55473	NA SECONDARY	20 mL	400 mg/L	0.04	1
STRONTIUM	55474	SR SECONDARY	10 mL	200 mg/L	0.02	1
THALLIUM	55475	TL SECONDARY	2.5 mL	5 mg/L	0.005	1
TIN	55476	SN SECONDARY	10 mL	20 mg/L	0.02	1
TITANIUM	55477	TI SECONDARY	5 mL	10 mg/L	0.01	1
Vanadium	55478	V SECONDARY	.5 mL	10 mg/L	0.001	1
ZINC	55479	ZN SECONDARY	1 mL	20 mg/L	0.002	1
Uranium	55480	U SECONDARY	1.25 mL	2.5 mg/L	0.0025	1

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Prepared Standard Data Sheet

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Serial #:	55693
Original Serial #:	
Identity:	SED SPK 4B
Description:	SED Spk 4B use 1ml/50ml samples for spk factor 1.0 / in 2% HNO3 Only
Type:	Intermediate
Group:	METALS
Status:	Active
Location:	B411
Storage Temp:	Room Temperature
Solvent Used:	Water
Prepped:	23-MAY-2014 12:00 by PETERSON_C
Expires:	30-MAY-2015 23:59
Entered:	27-MAY-14 by Charles J. Peterson
Discarded:	
Cancelled:	
Comments:	
Holding Time:	999 days, 0 hours
Final Volume:	500 mL
SOP Used for Preparation:	
Verified:	
Recertified:	

Table: Standard
Std/Reag Type:

See Also:

Component	Parent S/N	Parent Identity:	Amt of Parent Added	Final Conc	Dilution Factor	Mol Wt Factor
Sb	55454	SB SECONDARY	10 mL	20 mg/L	0.02	1
Silver	55472	AG SECONDARY	1 mL	2 mg/L	0.002	1

Appendix B

Appendix of Significant Changes

02/10/2009	Edited section 6.3.1 to clarify that the laboratory will not remove anything from the sample during preparation.
01/14/2010	Edited section 6.5 to say that we use reflux caps instead of watch glasses during sample reflux. Note about solid matrix waste samples needing to be digested with an LFB containing Ottawa Sand to represent matrix added to Section 8.4, point 2.
12/10/2014	Section 4 edited for more detail. Section 6 edited to include procedure for the new Prep Worksheet Manager. Appendices added.

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